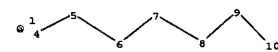
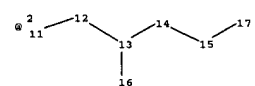
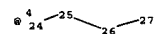
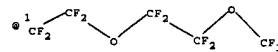
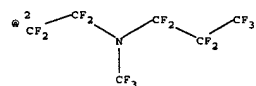
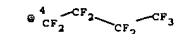
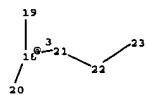
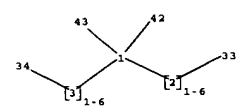
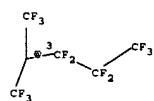
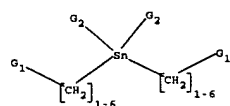


C:\STNEXP4\QUERIES\602105.str



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25
26 27 33 34 42 43

chain bonds :

1-3 1-2 1-42 1-43 2-33 3-34 4-5 5-6 6-7 7-8 8-9 9-10 11-12 12-13 13-14 13-16
14-15 15-17 18-19 18-20 18-21 21-22 22-23 24-25 25-26 26-27

exact/norm bonds :

1-42 1-43 2-33 3-34

exact bonds :

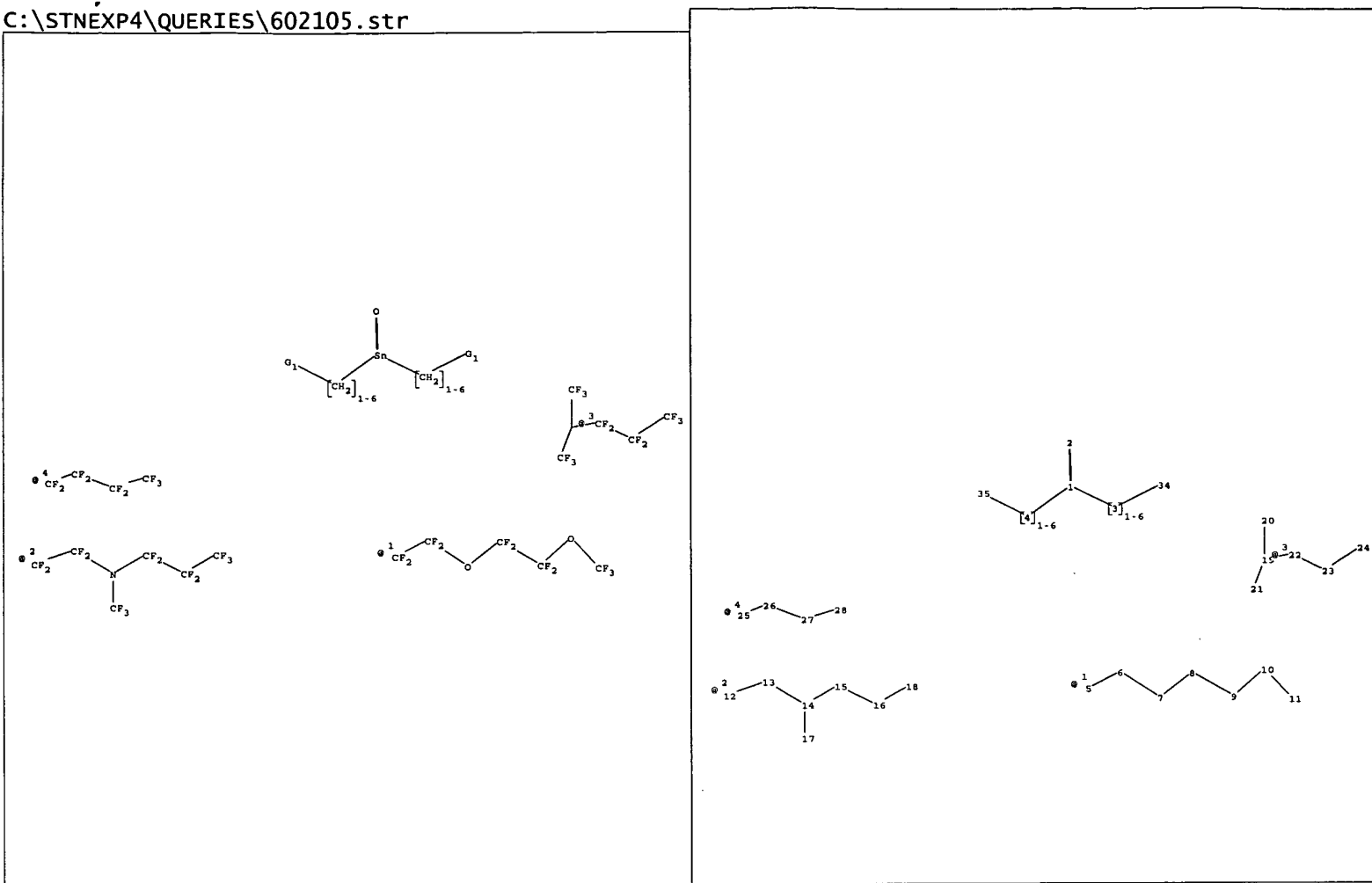
1-3 1-2 4-5 5-6 6-7 7-8 8-9 9-10 11-12 12-13 13-14 13-16 14-15 15-17 18-19
18-20 18-21 21-22 22-23 24-25 25-26 26-27

G1:CF3,[*1],[*2],[*3],[*4]

G2:X,Ak,Ph,OH,SH,MeO,EtO,n-PrO,i-PrO,n-BuO,i-BuO,s-BuO,t-BuO,PhO,CN,NH2

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS 10:CLASS
11:CLASS 12:CLASS 13:CLASS 14:CLASS 15:CLASS 16:CLASS 17:CLASS 18:CLASS 19:CLASS
20:CLASS 21:CLASS 22:CLASS 23:CLASS 24:CLASS 25:CLASS 26:CLASS 27:CLASS 33:CLASS
34:CLASS 42:CLASS 43:CLASS



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25
26 27 28 34 35

chain bonds :

1-2 1-3 1-4 3-34 4-35 5-6 6-7 7-8 8-9 9-10 10-11 12-13 13-14 14-15 14-17
15-16 16-18 19-20 19-21 19-22 22-23 23-24 25-26 26-27 27-28

exact/norm bonds :

3-34 4-35

exact bonds :

1-2 1-3 1-4 5-6 6-7 7-8 8-9 9-10 10-11 12-13 13-14 14-15 14-17 15-16 16-18
19-20 19-21 19-22 22-23 23-24 25-26 26-27 27-28

G1:CF3,[*1],[*2],[*3],[*4]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS 10:CLASS
11:CLASS 12:CLASS 13:CLASS 14:CLASS 15:CLASS 16:CLASS 17:CLASS 18:CLASS 19:CLASS
20:CLASS 21:CLASS 22:CLASS 23:CLASS 24:CLASS 25:CLASS 26:CLASS 27:CLASS 28:CLASS
34:CLASS 35:CLASS

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(FILE 'HOME' ENTERED AT 15:38:45 ON 29 SEP 2001)

FILE 'REGISTRY' ENTERED AT 15:39:00 ON 29 SEP 2001

```
L1          STRUCTURE UPLOADED
L2          QUE L1
L3          0 S L2
L4          0 S L2 FULL
L5          STRUCTURE UPLOADED
L6          QUE L5
```

=> d 15

L5 HAS NO ANSWERS

L5 STR

* STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT *

Structure attributes must be viewed using STN Express query preparation.

=> s 16 full

FULL SEARCH INITIATED 15:43:14 FILE 'REGISTRY'

FULL SCREEN SEARCH COMPLETED - 1220 TO ITERATE

100.0% PROCESSED 1220 ITERATIONS

0 ANSWERS

SEARCH TIME: 00.00.04

L7 0 SEA SSS FUL L5

09/602,105

Trying 3106016892...Open

Welcome to STN International! Enter x:x
LOGINID:sssptau124pxn
PASSWORD:
TERMINAL (ENTER 1, 2, 3, OR ?):2

* * * * * welcome to STN International * * * * *

NEWS 1 Web Page URLs for STN Seminar Schedule - N. America
NEWS 2 Dec 17 The CA Lexicon available in the CAPLUS and CA files
NEWS 3 Feb 06 Engineering Information Encompass files have new names
NEWS 4 Feb 16 TOXLINE no longer being updated
NEWS 5 Apr 23 Search Derwent WPINDEX by chemical structure
NEWS 6 Apr 23 PRE-1967 REFERENCES NOW SEARCHABLE IN CAPLUS AND CA
NEWS 7 May 07 DGENE Reload
NEWS 8 Jun 20 Published patent applications (A1) are now in USPATFULL
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DWPI and DPCI
NEWS 10 Aug 23 In-process records and more frequent updates now in
MEDLINE
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NEWS 13 Sep 17 IMSworld Pharmaceutical Company Directory name change
to PHARMASEARCH

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AND CURRENT DISCOVER FILE IS DATED 07 AUGUST 2001
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FILE 'HOME' ENTERED AT 15:50:53 ON 29 SEP 2001

=> fil reg

COST IN U.S. DOLLARS

SINCE FILE
ENTRY
0.15

TOTAL
SESSION
0.15

FULL ESTIMATED COST

FILE 'REGISTRY' ENTERED AT 15:51:06 ON 29 SEP 2001

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conducting SmartSELECT searches.

Structure search limits have been increased. See HELP SLIMIT
for details.

=>Testing the current file.... screen

ENTER SCREEN EXPRESSION OR (END):end

=>
Uploading C:\STNEXP4\QUERIES\602105.str

L1 STRUCTURE UPLOADED

=> que L1

L2 QUE L1

=> d his

(FILE 'HOME' ENTERED AT 15:50:53 ON 29 SEP 2001)

FILE 'REGISTRY' ENTERED AT 15:51:06 ON 29 SEP 2001

L1 STRUCTURE UPLOADED
L2 QUE L1

=> s 12 full

FULL SEARCH INITIATED 15:52:28 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 1723 TO ITERATE

100.0% PROCESSED 1723 ITERATIONS 18 ANSWERS
SEARCH TIME: 00.00.11

L3 18 SEA SSS FUL L1

=> fil caplus

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	134.18	134.33

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FILE COVERS 1947 - 29 Sep 2001 VOL 135 ISS 15
FILE LAST UPDATED: 28 Sep 2001 (20010928/ED)

09/602,105

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=> s 13
L4 9 L3

=> d 1-9 bib abs

L4 ANSWER 1 OF 9 CAPLUS COPYRIGHT 2001 ACS
AN 2001:27660 CAPLUS
DN 134:242980
TI Semi-Fluorinated Trialkyltin Fluorides and Fluorinated Telechelic Ionomers
as Viscosity-Enhancing Agents for Carbon Dioxide
AU Shi, Chunmei; Huang, Zhihua; Beckman, Eric J.; Enick, Robert M.; Kim, Sun-Young; Curran, Dennis P.
CS Department of Chemical and Petroleum Engineering, University of Pittsburgh, Pittsburgh, PA, 15261, USA
SO Ind. Eng. Chem. Res. (2001), 40(3), 908-913
CODEN: IECRED; ISSN: 0888-5885
PB American Chemical Society
DT Journal
LA English
AB Direct thickeners for dense carbon dioxide were designed and synthesized. Each thickener contained "CO₂-philic" fluorinated groups to impart soly. in carbon dioxide and "CO₂-phobic" functionalities to promote intermol. assocns. for viscosity enhancement. Semi-fluorinated trialkyltin fluorides and fluorinated telechelic ionomers were sol. to at least several wt. percent in dense liq. carbon dioxide without the use of a cosolvent. Increases in soln. viscosity at 297 K were measured using falling cylinder viscometry. The viscosity of liq. carbon dioxide was increased by a factor of 2-3 at thickener concns. of 2-4 wt. %. These results demonstrate that carbon dioxide viscosity enhancement is possible without the need for a cosolvent through the design of compds. with the appropriate balance of CO₂-philic groups for soly. and CO₂-phobic assocg. groups for macromol., viscosity-enhancing interactions. Neither compd., however, was as effective as the (29% styrene-71% fluoroacrylate) copolymer we recently developed. More substantial increases in soln. viscosity were not attained with the semi-fluorinated trialkyltin fluoride
fluoride because the fluorinated alkyl chains disrupted the assocns. that formed viscosity-enhancing, weakly assocg., linear polymers. The viscosity

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increases obtained with the telechelic ionomer were also less than expected because of the relatively low mol. wt. of the carbon-dioxide-sol.

ionomers. Higher-mol.-wt. ionomers would not be CO₂-sol., however.

RE.CNT 29

RE

(2) Cohen, L; US 3539311 1970 CAPLUS

(3) Curran, D; J Am Chem Soc 1999, V121(28), P6607 CAPLUS

(7) Enick, R; J Supercrit Fluids 1998, V13, P127 CAPLUS

(8) Guan, Z; Macromolecules 1994, V27, P5527 CAPLUS

(11) Harris, T; US 4913235 1990 CAPLUS

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 2 OF 9 CAPLUS COPYRIGHT 2001 ACS

AN 1999:410766 CAPLUS

DN 131:185032

TI Fluorous Tin Hydrides: A New Family of Reagents for Use and Reuse in Radical Reactions

AU Curran, Dennis P.; Hadida, Sabine; Kim, Sun-Young; Luo, Zhiyong
CS Department of Chemistry and Center for Combinatorial Chemist, University of Pittsburgh, Pittsburgh, PA, 15260, USA

SO J. Am. Chem. Soc. (1999), 121(28), 6607-6615

CODEN: JACSAT; ISSN: 0002-7863

PB American Chemical Society

DT Journal

LA English

OS CASREACT 131:185032

AB Eight members of a new family of highly fluorinated (fluorous) tin hydrides have been synthesized and studied as reagents for radical reactions. Tin hydrides of the general formulas $[Rf(CH_2)_n]_3SnH$ and $[Rf(CH_2)_n]Me_2SnH$ ($Rf = C_4F_9, C_6F_{13}, C_8F_{17}, C_{10}F_{21}$; $n = 2, 3$) were prepd. These reagents are highly sol. in fluorinated solvents, and partition coeffs. between perfluorohexanes and several org. solvents have been measured. The reagents are generally useful for reductive radical reactions and hydrostannation reactions that would typically be conducted with tributyltin hydride. Stoichiometric and catalytic procedures have been developed, and both feature very easy sepn. of the tin products from the org. products by convenient liq.-liq. or solid-liq. extns. The tin reagents are recovered from reactions in high yields and are routinely reused. Rate const. measurements suggest that the fluorous tin hydrides are about as reactive as (or in some cases, slightly more reactive than) tributyltin hydride. The reagents show excellent potential for large-scale application in "green" (environmentally friendly) processes. In addn., they are useful for combinatorial and parallel synthesis applications both as reagents and as scavengers in phase-switching procedures.

RE.CNT 81

RE

(2) Chatgililoglu, C; Acc Chem Res 1992, V25, P188 CAPLUS

(3) Chatgililoglu, C; J Org Chem 1995, V60, P3826 CAPLUS

(4) Clive, D; J Org Chem 1995, V60, P2607 CAPLUS

(5) Cornils, B; Angew Chem Int Ed Engl 1997, V36, P2057 CAPLUS

(6) Crich, D; J Org Chem 1996, V61, P7200 CAPLUS

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 3 OF 9 CAPLUS COPYRIGHT 2001 ACS

AN 1998:727787 CAPLUS

DN 130:66236

TI Rapid, parallel synthesis of homoallylic alcohols by Lewis acid mediated allylations of aldehydes with new fluorous allyl stannanes

AU Curran, Dennis P.; Luo, Zhiyong

CS Department of Chemistry, University of Pittsburgh, Pittsburgh, PA, 15260, USA
 SO Med. Chem. Res. (1998), 8(4/5), 261-265
 CODEN: MCREEB; ISSN: 1054-2523
 PB Birkhaeuser Boston
 DT Journal
 LA English
 AB Parallel Lewis acid mediated allylations of four arylaldehydes with two fluororous allylstannanes are reported. These stannanes tris-(4,4,5,5,6,6,7,7,8,8,9,9-tridecafluorononyl)allylstannane and tris-(4,4,5,5,6,6,7,7,7-nonafluoroheptyl)allylstannane bear propylene spacers between the fluoroalkyl groups and the Sn and seem to behave like normal trialkylstannanes in SnCl₄ promoted allylations. Reactions are purified by quenching with base and filtration through fluororous reverse phase silica gel. The procedure is prototypical of a general method to use fluororous reagents for conducting ionic reactions in parallel.

RE.CNT 21

RE

- (1) Cornils, B; Angew Chem Int Ed 1997, V36, P2057 CAPLUS
 (4) Curran, D; Chemtracts-Org Chem 1996, V9, P75 CAPLUS
 (5) Curran, D; J Am Chem Soc 1996, V118, P2531 CAPLUS
 (6) Curran, D; J Org Chem 1996, V61, P6480 CAPLUS
 (7) Curran, D; J Org Chem 1997, V62, P6714 CAPLUS

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 4 OF 9 CAPLUS COPYRIGHT 2001 ACS

AN 1998:630478 CAPLUS

DN 129:330310

TI "Propylene spaced" allyl tin reagents: a new class of fluororous tin reagents for allylations under radical and metal-catalyzed conditions

AU Curran, Dennis P.; Luo, Zhiyong; Degenkolb, Peter

CS Department Chemistry, University Pittsburgh, Pittsburgh, PA, 15260, USA

SO Bioorg. Med. Chem. Lett. (1998), 8(17), 2403-2408

CODEN: BMCLE8; ISSN: 0960-894X

PB Elsevier Science Ltd.

DT Journal

LA English

AB A new generation of propylene-spaced fluororous allyltin reagents [(Rf(CH₂)₃)₃SnCH₂CH=CH₂] [Rf = CF₃(CF₂)_n(CH₂)₃; n = 5, 3] is described. These succeed in radical allylations where their lower homologs (ethylene-spaced) fail, and they provide improved performance in transition metal catalyzed allylations. The reagents and byproducts are readily sepd. by simple fluororous-org. liq.-liq. or solid-liq. extns.

L4 ANSWER 5 OF 9 CAPLUS COPYRIGHT 2001 ACS

AN 1983:595223 CAPLUS

DN 99:195223

TI Tris-(.gamma.-trifluoropropyl)chlorostannane

IN Mironov, V. F.; Yankov, V. V.; Stepina, E. M.; Kuptsova, T. S.; Shiryaev, V. I.

PA USSR

SO U.S.S.R.

From: Otkrytiya, Izobret., Prom. Obraztsy, Tovarnye Znaki 1983, (24), 206.

CODEN: URXXAF

DT Patent

LA Russian

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	SU 536680	A1	19830630	SU 1975-2137216	19750526

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AB (F3CCH2CH2)3SnCl was prepd. by treating F3CCH2CH2Cl with Sn (3-10:1 molar ratio) at 100-200.degree. in the presence of amine or phosphine catalyst (0.1-0.5 mol/mol Sn) and iodine or iodine-contg. compd. cocatalyst (0.05-0.2 mol/mol Sn).

L4 ANSWER 6 OF 9 CAPLUS COPYRIGHT 2001 ACS

AN 1978:509698 CAPLUS

DN 89:109698

TI Direct preparation of organotin biocides

AU Stepina, E. M.; Yankov, V. V.; Gulo, R. A.; Kuptsova, T. S.; Mironov, V. F.

CS Minist. Khim. Prom., Moscow, USSR

SO Biol. Akt. Soedin. Elem. IV B Gruppy (1977), 231-4 Publisher: Akad. Nauk SSSR, Sib. Otd., Irkutsk. Inst. Org. Khim., Irkutsk, USSR.

CODEN: 38OBA2

DT Conference

LA Russian

AB (F3CCH2CH2)3SnCl was prepd. in 80% yields by heating F3CCH2CH2Cl with Sn in presence of amines and iodine at 170-80.degree.. Test data for its antiseptic properties were given.

L4 ANSWER 7 OF 9 CAPLUS COPYRIGHT 2001 ACS

AN 1971:488765 CAPLUS

DN 75:88765

TI Fluoroalkyltin compounds

IN Murch, Robert M.

PA Dow Corning Corp.

SO U.S., 2 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 3590060	A	19710629	US 1969-860028	19690922
AB	3,3,3-Trifluoropropyltin compds. are prepd. for use as stabilizers in polyvinyl resins, catalysts, pesticides, bactericides, and oil repellents.				
	CF3(CH2)2MgBr was treated with SnCl4 to give tetrakis(3,3,3 - trifluoropropyl)tin. (3,3,3 - Trifluoropropyl)triphenyltin and tris(3,3,3-trifluoropropyl)methoxytin were among 9 compds. similarly prepd.				

L4 ANSWER 8 OF 9 CAPLUS COPYRIGHT 2001 ACS

AN 1970:409206 CAPLUS

DN 73:9206

TI Nuclear magnetic resonance coupling constants in tin in 3,3,3-trifluoropropyltin compounds

AU Williams, Dwight Edward; Toporcer, Louis H.; Ronk, Gary M.

CS Dow Corning Corp., Midland, Mich., USA

SO J. Phys. Chem. (1970), 74(10), 2139-42

CODEN: JPCHAX

DT Journal

LA English

AB The various NMR coupling consts. between Sn and H and F have been detd. from the heteronuclearly decoupled ¹H and ¹⁹F NMR spectra of eleven 3,3,3-trifluoropropyltin compds. These results are discussed in the

light

of the Barfield alternate MO theory of spin coupling. The invalidity of the use of coupling consts. to det. hybridization is briefly discussed. For these data it was not necessary to invoke a "spatial mechanism" of

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spin coupling. The vicinal Sn-H coupling const. was larger than the geminal Sn-H coupling const. in most instances despite reports that the opposite case is a characteristic of metal-alkyl compds. These data indicate that a postulated intramol. interaction between Sn and F does not occur.

L4 ANSWER 9 OF 9 CAPLUS COPYRIGHT 2001 ACS
AN 1970:72974 CAPLUS
DN 72:72974

TI Orbital populations and π backbonding in some organohalostannanes: interpretation of tin Moessbauer and ligand NQR[nuclear quadrupole resonance] data

AU Williams, Dwight Edward; Kocher, C. W.
CS Dow Corning Corp., Midland, Mich., USA
SO J. Chem. Phys. (1970), 52(3), 1480-8
CODEN: JCPSA6

DT Journal
LA English

AB Tin Moessbauer isomer shifts and quadrupole splittings are reported for
23

compds. Equations are derived and applied which enable the Sn hybrid orbital populations for tetravalent Sn compds. of the type AmSnB_{4-m} to be detd. from the above data. Crit. tests of the theory are proposed. A min. value of 5.8 mm sec⁻¹ is obtained for the quadrupole splitting due

to one p electron by comparing Moessbauer-derived with ligand NQR-derived populations. The two sets of populations are also used to obtain π backbonding orbital populations in organohalostannanes. The Sn sp^3 orbitals are more important than d orbitals for π back-bonding.